

## SYNTHESIS OF NEW HETEROCYCLIC THIAZOLIDINONE AND AZETIDINONE COMPOUNDS AND THEIR ANTICANCER ACTIVITY

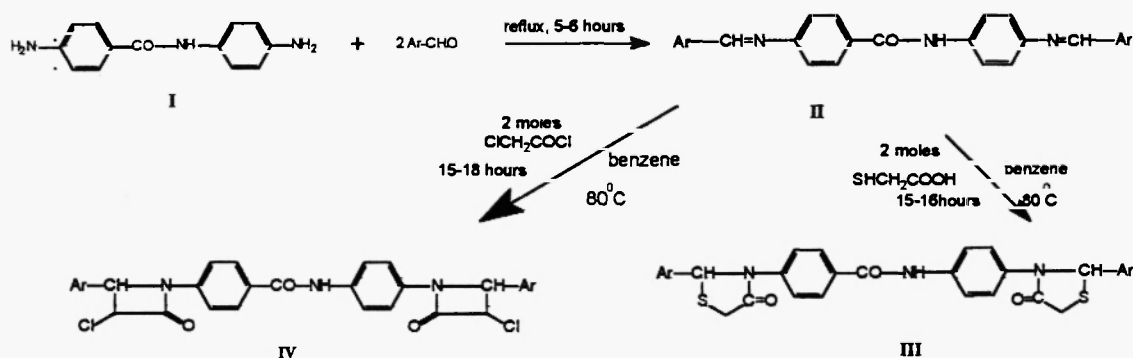
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**Abstract** :- Thiazolidinones and azetidinones have been prepared by the reaction of various schiff bases with thioglycolic acid and chloroacetyl chloride respectively. The intermediate schiff bases were synthesised by the condensation of diaminobenzanilide with various aldehydes. The thiazolidinones and azetidinones were tested for their anticancer activity.

### Introduction

Diaminobenzanilide was found to exhibit insecticidal property<sup>1</sup>, growth inhibition of plants<sup>2</sup>, tuberculostatic action<sup>3</sup> and herbicidal<sup>4</sup> property. Thiazolidinones are known to exhibit antitubercular, antibacterial<sup>5,6</sup>, anticonvulsant<sup>7,8</sup>, antifungal<sup>9</sup>, antithyroid and amoebicidal<sup>10</sup> activities. Azetidinones ( $\beta$ -lactams) were tested as antibiotics, antidepressants, sedatives<sup>11</sup>. So an attempt was made to synthesize some thiazolidinones and azetidinones using diaminobenzanilide as the starting material and test them as anti-cancer drug. Diaminobenzanilide (I) was condensed with different aromatic aldehydes to yield di-imines (II) (schiff base).



The di-imines (schiff base, II) were further reacted with thioglycolic acid & chloroacetyl chloride to yield thiazolidinones (III) & azetidinones (IV) respectively.

The starting compound, diaminobenzanilide, shows IR absorption peak at  $3400\text{--}3300\text{ cm}^{-1}$  and  $3200\text{--}3100\text{ cm}^{-1}$ ,  $1560\text{ cm}^{-1}$  (C-N stretching and bending for amino group),  $1700\text{--}1660\text{ cm}^{-1}$  (C=O stretching). The schiff bases of above starting compound shows IR absorption peak at  $1587\text{--}1548\text{ cm}^{-1}$  (C=N stretching),  $1660\text{--}1680\text{ cm}^{-1}$  (C=O stretching). The thiazolidinone compounds were characterized by their IR absorption bands at  $3330\text{--}3300\text{ cm}^{-1}$  (N-H stretching),  $720\text{--}600\text{ cm}^{-1}$  (C-S stretching),  $1750\text{--}1680\text{ cm}^{-1}$  (C=O stretching) and  $1590\text{--}1560\text{ cm}^{-1}$  (C-N stretching). The azetidinone compounds were characterized by their IR absorption bands at  $1730\text{--}1680\text{ cm}^{-1}$  (C=O stretching),  $1715\text{ cm}^{-1}$  and  $730\text{ cm}^{-1}$  (C-Cl stretching and bending).

**Experimental**

All the melting points are taken in an open capillary and are uncorrected. The IR spectra was recorded with KBr pellets on Perkin-Elmer 783 Spectrophotometer.

**Preparation of Schiff bases<sup>12</sup>**

Diaminobenzanilide (0.02 mol, 4.54 g) was taken in benzene in a Dean-Stark apparatus and to it benzaldehyde (0.04 mol, 4.24 g) was added over a period of 15 minutes. Then the mixture was refluxed for 5 hours. During the course of the reaction the water was removed continuously. The benzene was then distilled off to get the product. The Schiff's base was recrystallised in benzene. Other substituted Schiff bases were prepared in a similar manner. The analytical data for various substituted schiff bases are given in Table-1

**Preparation of Thiazolidinone<sup>13</sup>**

The schiff's base (0.0075 mol, 3.0 g) in benzene was taken in Dean-Stark apparatus. To it thioglycolic acid (0.015 mol, 1.37 g) in benzene was added slowly. Then it was refluxed for 15-16 hours. During the course of the reaction the water was removed continuously. The benzene was distilled off to get the product. Other substituted thiazolidinones were prepared in similar manner. The analytical data for different substituted thiazolidinones are given in Table-2

**Preparation of Azetidinone<sup>14</sup>**

The schiff's base (0.0075 mol, 3.0 g) in benzene was taken in a 50 ml flat bottomed flask. To it chloroacetyl chloride (0.015 mol, 1.68 g) and triethyl amine (0.015 mol, 1.50 g) in benzene were added slowly. It was then refluxed for 15-16 hours. The triethylamine hydrochloride was removed and the benzene was distilled off to get the product. Other substituted azetidinones were prepared in a similar manner. The analytical data for different substituted azetidinones are given in Table-3.

**Report on the anti-cancer activity.****Experimental**

The compounds were also screened for their anti-cancer activity by measuring their effect on percentage growth (PG) of more than two different cell lines for variety of cancer. They have been tested at 5 different concentration of the compound ( $-4 \log_{10}$  to  $-8 \log_{10}$ ). The optical density of SRB derived colour by the cell lines was measured at 0 time ( $\text{Mean}_{\text{zero}}$ ) after 48 hours in presence of drug ( $\text{Mean}_{\text{test}}$ ) and in absence of drug after 48 hours ( $\text{Mean}_{\text{control}}$ ). The PG was calculated from it using the following formula-

If  $\text{Mean}_{\text{test}} - \text{Mean}_{\text{zero}} > 0$  then,

$$\text{PG} = \frac{100 \times (\text{Mean}_{\text{test}} - \text{Mean}_{\text{zero}})}{(\text{Mean}_{\text{control}} - \text{Mean}_{\text{zero}})}$$

But if  $(\text{Mean}_{\text{test}} - \text{Mean}_{\text{zero}}) < 0$  then,

$$\text{PG} = \frac{100 \times (\text{Mean}_{\text{test}} - \text{Mean}_{\text{zero}})}{\text{Mean}_{\text{zero}}}$$

The effects were interpreted from dose response curves created by plotting PG's (-100 to +100) against  $\log_{10}$  molar concentration (-4 to -8). The response parameter GI50, TGI and LC50 are interpolated values representing the concentrations at which the PG is +50, 0 and -50.

**Results and Discussion**

The invitro effect of compound number-16, 4,4'-bis[2"-oxo-5"-(3"',4"',5"'-trimethoxyphenyl)thiazolidine] - benzanilide drug at different concentrations expressed as percentage growth (PG). From it three response parameters GI50, TGI (Test Growth Index) and LC50 (Lethal Concentration showing -50 PG) are interpolated. The compounds for which it was not possible to determine their response parameters were tested at highest concentration. Only in some of the cell lines the compounds were found to be effective at  $-4 \log_{10}$  M concentration. Similarly from the invitro effect for compound number-12, 20 and 15 shows that compound-12 is most effective among the three of them showing nearly -25PG.

The dose of the effective compounds were determined from the Dose Response Curves. All 56 cell lines tested for different panel, i.e. organ cancer were effective at  $-4\log 10M$  concentration while higher concentrations (i.e. -5, -6, -7 and  $-8\log 10M$ ) were comparatively less effective. It was found that out of 56 cell lines only 5 viz. HOP-92 of lung cancer, UACC-62 of Melanoma and NCI/ADR-RES, MDA-N and MDA-MB-435 of Breast Cancer were effective as their TGI is -4.10, -4.12, -4.15, -4.19 and -4.21 respectively. The compound is most effective for MDA-MB-435 cell line of Breast Cancer.

For selecting a potential drug for a particular cell line or for a particular cancer, mean graphs have been obtained for all the three interpolated values G150, TGI and LC50 for compound 16, 12, 20 and 15 respectively. Similarly mean graphs were also obtained for compound-12, 4,4'-bis[2"-oxo-5"-(2"-hydroxyphenyl)thiazolidine]benzanilide, compound-20, 4,4'-bis[2"-oxo-5"-(4"-N,N-dimethyl phenyl)thiazolidine]benzanilide and compound-15, 4,4'-bis[2"-oxo-5"-(thiophenyl)thiazolidine]benzanilide, against all 56 cell lines. Among them compound-12, 4,4'-bis[2"-oxo-5"-(2"-hydroxyphenyl)thiazolidine]benzanilide, is effective on most of the cell lines at  $>-4\log 10M$  concentration. It was found that out of 56 cell lines tested only five were showing its effectiveness on the cell lines viz. HOP-92 for Lung Cancer, SNB-75 of CNS cancer, M-14, UACC-62 for Melanoma and SK-OV-3 for Ovarian Cancer having TGI -4.06, -4.08, -4.21 and -4.21 respectively. It was found that this compound shows  $>-4.0$  TGI only in case of four cell lines. Among four of them M-14 and SK-OV-3 were affected by M1

Compound-20, 4,4'-bis[2"-oxo-5"-(4"-N,N-dimethylphenyl)thiazolidine]benzanilide, is totally ineffective showing TGI -4.0. Similarly compound-15, 4,4'-bis[2"-oxo-5"-(thiophenyl)thiazolidine]benzanilide, is effective only on one cell line i.e. NCI-H23 for Lung Cancer having TGI -4.02. The rest of the 55 cell lines show TGI -4.0 at  $-4\log 10M$  concentration.

Finally we can conclude that all compounds were effective on lung cancer except compound-20, but on different cell lines. Two of them, compound-16 and 12 are effective on UACC-62 cell line for Melanoma. It was specifically found that compound-16 is effective on 3 different cell lines for Breast Cancer. The compound-12 is also effective on Ovarian Cancer cell line SK-OV-3. Among the four compound-12 is most effective on Melanoma Cancer M-14 cell lines and SK-OV-3 for Ovarian Cancer showing  $-4.21\log 10M$  and compound-12 on Breast Cancer cell line MDA-MB-435 showing nearly similar TGI. The latter anti-cancer agent also shows its effectiveness, but slightly less, on MDA-N, another cell line for Breast Cancer as the log TGI for it is -4.19. None of the four compound is showing  $-50PG$ .

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Table-1 THE ANALYTICAL DATA OF THE SCHIFF'S BASE

| Compound Number | Substituent Ar-  | Molecular Formula  | Molecular Weight(gms) | Melting Point °C | % Yield | Elemental Analysis, %Theoretical (Practical) |            |              |
|-----------------|--|--|-----------------------|------------------|---------|--|------------|--------------|
|                 |  |  |                       |                  |         | C  | H          | N            |
| 1               | C <sub>6</sub> H <sub>5</sub> -  | C <sub>27</sub> H <sub>21</sub> N <sub>3</sub> O               | 403                   | 192              | 91      | 80.39(80.41)                                 | 5.21(5.24) | 10.42(10.44) |
| 2               | 4-OCH <sub>3</sub> ,C <sub>6</sub> H <sub>4</sub> -                    | C <sub>29</sub> H <sub>23</sub> N <sub>3</sub> O <sub>3</sub>  | 463                   | 176              | 88      | 69.97(69.95)                                 | 5.39(5.36) | 9.07(9.09)   |
| 3               | 2-OH,C <sub>6</sub> H <sub>4</sub> -                                   | C <sub>27</sub> H <sub>21</sub> N <sub>3</sub> O <sub>3</sub>  | 435                   | 210              | 95      | 74.48(74.44)                                 | 4.82(4.84) | 9.65(9.67)   |
| 4               | C <sub>6</sub> H <sub>5</sub> -CH=CH-                                  | C <sub>31</sub> H <sub>23</sub> N <sub>3</sub> O               | 455                   | 188              | 93      | 81.75(81.77)                                 | 5.49(5.47) | 9.23(9.21)   |
| 5               | C <sub>4</sub> H <sub>9</sub> S-                                       | C <sub>23</sub> H <sub>17</sub> N <sub>3</sub> OS <sub>2</sub> | 415                   | 132              | 89      | 66.50(66.51)                                 | 4.09(4.11) | 10.12(10.14) |
| 6               | 3,4,5-(OCH <sub>3</sub> ) <sub>3</sub> C <sub>6</sub> H <sub>2</sub> - | C <sub>33</sub> H <sub>33</sub> N <sub>3</sub> O <sub>7</sub>  | 583                   | 183              | 95      | 67.92(67.94)                                 | 5.66(5.65) | 7.20(7.22)   |
| 7               | 2-NO <sub>2</sub> ,C <sub>6</sub> H <sub>4</sub> -                     | C <sub>27</sub> H <sub>19</sub> N <sub>3</sub> SO <sub>3</sub> | 493                   | 175              | 75      | 65.72(65.75)                                 | 3.85(3.86) | 14.19(14.18) |
| 8               | 4-Cl,C <sub>6</sub> H <sub>4</sub> -                                   | C <sub>27</sub> H <sub>19</sub> N <sub>3</sub> OC <sub>2</sub> | 472                   | 141              | 82      | 68.64(68.66)                                 | 4.02(4.05) | 8.89(8.87)   |
| 9               | 3-OC <sub>2</sub> H <sub>5</sub> ,4-OH,C <sub>6</sub> H <sub>3</sub> - | C <sub>31</sub> H <sub>29</sub> N <sub>3</sub> O <sub>3</sub>  | 523                   | 164              | 78      | 71.12(71.13)                                 | 5.54(5.57) | 8.03(8.05)   |
| 10              | N,N-(CH <sub>3</sub> ) <sub>2</sub> ,C <sub>6</sub> H <sub>4</sub> -   | C <sub>29</sub> H <sub>23</sub> N <sub>3</sub> O               | 459                   | 180              | 93      | 75.81(75.83)                                 | 5.54(5.53) | 15.25(15.22) |

Table-2 THE ANALYTICAL DATA OF THE THIAZOLIDINONES

| Compound Number | Substituent Ar-  | Molecular Formulac   | Molecular Weight(gms) | Melting Point °C | % Yield | Elemental Analysis, %Theoretical (Practical) |            |              |
|-----------------|--|--|-----------------------|------------------|---------|--|------------|--------------|
|                 |  |  |                       |                  |         | C  | H          | N            |
| 11              | C <sub>6</sub> H <sub>5</sub> -  | C <sub>33</sub> H <sub>23</sub> N <sub>3</sub> O <sub>3</sub> S <sub>2</sub>                 | 551                   | 118              | 74      | 67.51(67.48)                                 | 4.53(4.50) | 7.62(7.65)   |
| 12              | 4-OCH <sub>3</sub> ,C <sub>6</sub> H <sub>4</sub> -                    | C <sub>33</sub> H <sub>29</sub> N <sub>3</sub> O <sub>3</sub> S <sub>2</sub>                 | 611                   | 141              | 78      | 64.81(64.76)                                 | 4.74(4.69) | 6.87 (6.80)  |
| 13              | 2-OH,C <sub>6</sub> H <sub>4</sub> -                                   | C <sub>31</sub> H <sub>23</sub> N <sub>3</sub> O <sub>3</sub> S <sub>2</sub>                 | 583                   | 159              | 85      | 63.80(63.84)                                 | 4.28(4.33) | 7.20(7.22)   |
| 14              | C <sub>6</sub> H <sub>5</sub> -CH=CH-                                  | C <sub>33</sub> H <sub>29</sub> N <sub>3</sub> O <sub>3</sub> S <sub>2</sub>                 | 603                   | 123              | 69      | 69.65(69.62)                                 | 4.80(4.83) | 6.97(6.99)   |
| 15              | C <sub>4</sub> H <sub>9</sub> S-                                       | C <sub>27</sub> H <sub>21</sub> N <sub>3</sub> O <sub>3</sub> S <sub>4</sub>                 | 563                   | 89               | 71      | 57.54(57.55)                                 | 3.73(3.76) | 7.46(7.44)   |
| 16              | 3,4,5-(OCH <sub>3</sub> ) <sub>3</sub> C <sub>6</sub> H <sub>2</sub> - | C <sub>37</sub> H <sub>27</sub> N <sub>3</sub> O <sub>6</sub> S <sub>2</sub>                 | 731                   | 112              | 85      | 60.73(60.75)                                 | 5.06(5.04) | 5.74(5.70)   |
| 17              | 2-NO <sub>2</sub> ,C <sub>6</sub> H <sub>4</sub> -                     | C <sub>31</sub> H <sub>23</sub> N <sub>3</sub> O <sub>5</sub> S <sub>2</sub>                 | 641                   | 131              | 73      | 58.03(58.07)                                 | 3.58(3.57) | 10.72(10.68) |
| 18              | 4-Cl,C <sub>6</sub> H <sub>4</sub> -                                   | C <sub>31</sub> H <sub>23</sub> N <sub>3</sub> O <sub>3</sub> Cl <sub>2</sub> S <sub>2</sub> | 620                   | 127              | 82      | 60.00(60.02)                                 | 3.70(3.73) | 6.77(6.73)   |
| 19              | 3-OC <sub>2</sub> H <sub>5</sub> ,4-OH,C <sub>6</sub> H <sub>3</sub> - | C <sub>35</sub> H <sub>33</sub> N <sub>3</sub> O <sub>7</sub> S <sub>2</sub>                 | 671                   | 114              | 78      | 62.59(62.57)                                 | 4.91(4.92) | 6.25(6.27)   |
| 20              | N,N-(CH <sub>3</sub> ) <sub>2</sub> ,C <sub>6</sub> H <sub>4</sub> -   | C <sub>33</sub> H <sub>29</sub> N <sub>3</sub> O <sub>3</sub> S <sub>2</sub>                 | 607                   | 105              | 63      | 65.23(65.20)                                 | 4.77(4.74) | 11.53(11.55) |

Table-3 **THE ANALYTICAL DATA OF THE AZETIDINONES**

| Compound Number | Substituent Ar-  | Molecular Formulac   | Molecular Weight(gms) | Melting Point °C | % Yield | Elemental Analysis, %Theoretical (Practical) |            |              |
|-----------------|--|--|-----------------------|------------------|---------|--|------------|--------------|
|                 |  |  |                       |                  |         | C  | H          | N            |
| 21              | C <sub>6</sub> H <sub>5</sub> -  | C <sub>31</sub> H <sub>23</sub> N <sub>3</sub> O <sub>3</sub> Cl <sub>2</sub>                | 556                   | >250             | 92      | 66.90(66.87)                                 | 4.13(4.11) | 7.55 (7.51)  |
| 22              | 4-OCH <sub>3</sub> , C <sub>6</sub> H <sub>4</sub>                       | C <sub>33</sub> H <sub>27</sub> N <sub>3</sub> O <sub>4</sub> Cl <sub>2</sub>                | 616                   | >250             | 78      | 64.28(64.26)                                 | 4.38(4.34) | 6.81(6.79)   |
| 23              | 2-OH, C <sub>6</sub> H <sub>4</sub> -                                    | C <sub>31</sub> H <sub>23</sub> N <sub>3</sub> O <sub>3</sub> Cl <sub>2</sub>                | 588                   | >250             | 89      | 63.26(63.28)                                 | 3.91(3.94) | 7.14(7.18)   |
| 24              | C <sub>6</sub> H <sub>5</sub> -CH=CH-                                    | C <sub>33</sub> H <sub>27</sub> N <sub>3</sub> O <sub>3</sub> Cl <sub>2</sub>                | 608                   | >250             | 95      | 69.07(69.05)                                 | 4.40(4.43) | 6.90(6.88)   |
| 25              | C <sub>6</sub> H <sub>5</sub> S-   | C <sub>27</sub> H <sub>19</sub> N <sub>3</sub> O <sub>3</sub> S <sub>2</sub> Cl <sub>2</sub> | 568                   | >250             | 86      | 57.04(57.06)                                 | 3.34(3.37) | 7.39(7.41)   |
| 26              | 3,4,5-(OCH <sub>3</sub> ) <sub>3</sub> , C <sub>6</sub> H <sub>2</sub> - | C <sub>37</sub> H <sub>33</sub> N <sub>3</sub> O <sub>6</sub> Cl <sub>2</sub>                | 736                   | >250             | 91      | 60.32(60.29)                                 | 4.75(4.73) | 5.70(5.71)   |
| 27              | 2-NO <sub>2</sub> , C <sub>6</sub> H <sub>4</sub> -                      | C <sub>31</sub> H <sub>21</sub> N <sub>3</sub> O <sub>7</sub> Cl <sub>2</sub>                | 646                   | >250             | 78      | 57.58(57.55)                                 | 3.25(3.27) | 10.83(10.86) |
| 28              | 4-Cl, C <sub>6</sub> H <sub>4</sub> -                                    | C <sub>31</sub> H <sub>21</sub> N <sub>3</sub> O <sub>3</sub> Cl <sub>4</sub>                | 625                   | >250             | 92      | 59.52(59.49)                                 | 3.36(3.33) | 6.72(6.71)   |
| 29              | 3-OC <sub>2</sub> H <sub>5</sub> , 4-OH, C <sub>6</sub> H <sub>3</sub>   | C <sub>33</sub> H <sub>31</sub> N <sub>3</sub> O <sub>7</sub> Cl <sub>2</sub>                | 676                   | >250             | 84      | 61.58(61.56)                                 | 5.42(5.39) | 6.15(6.11)   |
| 30              | N,N-(CH <sub>3</sub> ) <sub>2</sub> , C <sub>6</sub> H <sub>4</sub> -    | C <sub>33</sub> H <sub>27</sub> N <sub>4</sub> O <sub>3</sub> Cl <sub>2</sub>                | 612                   | >250             | 87      | 64.70(64.73)                                 | 4.41(4.44) | 11.43(11.45) |

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